NEODOLABELLINE, A METHYL MIGRATED DOLABELLANE-TYPE DITERPENE FROM THE OKINAWAN SOFT CORAL CLAVULARIA KOELLIKERI

- Motomasa Kobayashi,^{a)} Byeng Wha Son,^{a)} Takaji Fujiwara,^{a)} Yoshimasa Kyoqoku,^{b)} and Isao Kitagawa^{*a)}
- a) Faculty of Pharmaceutical Sciences, Osaka University, 1-6, Yamada-oka, Suita, Osaka 565, Japan
- b) Institute for Protein Research, Osaka University, 3-2, Yamada-oka, Suita, Osaka 565, Japan

Summary: A new diterpene named neodolabelline (5), having a methyl migrated dolabellane-type skeleton, was isolated from the stoloniferan soft coral *Clavularia koellikeri*, together with clavukerins (trinor-sesquiterpenes) and bicyclogermacrene, and the absolute configuration was determined.

In the course of chemical studies on the constituents of Okinawan marine organisms,¹⁾ we isolated three new trinor-sesquiterpenes named clavukerins A (1), B (2), and C (3) together with bicyclogermacrene (4) from the stolonifer *Clavularia koellikeri* and elucidated their absolute configurations.²⁾ In a continuing study of the same soft coral (collected in July 1983, at Kohamajima, Okinawa Prefecture), we isolated a new diterpene named neodolabelline (5), which possessed a methyl migrated dolabellane-type skeleton, and elucidated the absolute configuration as described in this communication.

An acetone extract of the fresh soft coral was partitioned into an AcOEtwater mixture and the AcOEt soluble portion was repeatedly chromatographed on SiO₂ to furnish neodolabelline (5) (7.5% from the AcOEt extract), clavukerins, and bicyclogermacrene.³⁾ The IR spectrum of neodolabelline (5), a colorless amorphous solid, $C_{20}H_{30}O_4$,⁴⁾ $[\alpha]_D^{19}$ +53° (CHCl₃), showed hydroxyl and carbonyl absorption bands [3595, 3410 (br), 1710 cm⁻¹], while the ¹³C NMR spectrum (22.5 MHz, d_{6} -DMSO), showed signals due to one carbonyl carbon [δc 208.3 (s), C-9], two olefinic carbons [147.7 (s), C-1; 124.4 (d), C-14], five carbons having an oxygen function [77.0 (s), C-8; 77.0 (d), C-3; 66.0 (d), C-7; 58.6 (s), C-4; 58.2 (d), C-5], four methylene carbons [38.6 (t), C-10; 34.1, 25.0, 23.4 (all t), C-2, 6, 13], two methine carbons [47.8 (d), C-12; 28.9 (d), C-18], one quaternary carbon [52.6 (s), C-11], five methyl carbons [22.8, 22.3, 21.6, 20.7, The ¹H NMR spectrum (500 MHz)⁵⁾ of 5 showed signals assignable 15.9 (all q)]. to two sec.methyls [δ 1.04, 0.90 (both 3H, d, J=6.5, 18-(CH₂)₂], three tert. methyls [1.26, 0.94, 0.85 (each 3H, s, 4, 8, 11-CH₃)], one olefinic proton [5.41 (1H, br s, 14-H)], three protons geminal each to an oxygen function [3.82 (1H, dd, J=11.5, 3.5, 3-H), 3.29 (1H, br s, 5-H), 3.97 (1H, d, J=10.5, 7-H)] and the decoupling experiments in detail revealed that 5 was a tetracyclic

diterpene.

Deuteration (NaOCH₃-CH₃OD) of neodolabelline (5) resulted in the loss of signals due to $10-H_2$ and C-10 (¹H, ¹³C NMR), and acetylation of 5 provided the The comparison of ¹³C NMR data monoacetate ($\underline{6}$), $C_{22}H_{32}O_5$ (δ 2.03, 3H, s). for 5 and 6 led us to presume the presence of an α -ketol moiety [ma-CH₂-CO-C(OH) (CH_3) -] in 5. NaBH₄ reduction of 5 yielded a trans-diol (7)(45%), $C_{20}H_{32}O_4$ (δ 3.94, 1H, br s, 9-H) and a *cis*-diol (8)(15%), $C_{20}H_{32}O_4$ (3.95, 1H, d, J=9.0, Upon irradiation of the 8-CH₃ signal of <u>8</u>, 7% NOE was observed on the 9-H). NaIO₄ oxidation of $\frac{7}{2}$ and $\frac{8}{2}$ afforded quantitatively an identical 9-H signal. keto-aldehyde (9), $C_{20}H_{30}O_4$ (8 9.68, 1H, dd, J=4.0, 2.0, 9-H; 2.14, 3H, s, 8-CH3). NaBH4 reduction followed by acetylation of 9 yielded a pair of epimeric diacetates: 10, C₂₄H₃₈O₆ [δ 3.50 (1H, ddd, J=10.5, 6.0, 2.5, 7-H), 4.77 (1H, dq, J=6.0, 6.0, 8-H), 4.13 (1H, ddd, J=10.5, 10.0, 6.5, 9-Ha), 3.89 (1H, ddd, J=10.5, 10.5, 5.0, 9-Hb), 2.04, 2.03 (both 3H, s, 8, 9-OAc)], and 8-epimer(11), $C_{24}H_{38}O_6$. Thus, the partial structure of 5 from C-6 to C-10 (22-CH₂-CO-C(OH) (CH₂)-CH(O-)-CH₂-] was elucidated.

Treatment of neodolabelline (5) with 47% aq. HBr cleaved the 4,5-epoxide ring to furnish two bromohydrins: 12, $C_{20}H_{31}O_4Br$ [δ 4.64 (1H, dd, J=11.5, 4.5, 5-H), 2.20 (1H, br dd, J=12.5, 4.5, 6 α -H), 1.75 (1H, m, 6 β -H), 1.29 (3H, s, 4-CH₃)], and 13, $C_{20}H_{31}O_4Br$ [δ 4.11 (1H, br s, 5-H), 2.09 (1H, dd, J=13.0, 11.5, 6 α -H), 1.61 (1H, dd-like, J=13.0, 2.0, 6 β -H), 1.78 (3H, s, 4-CH₃)]. Thus, the elucidated partial structure of 5 was extended from C-4 to C-10.

NBS treatment with irradiation⁶⁾ of neodolabelline (5) in 90% aq. dioxane provided 1,14-dibromo-neodolabelline (14), mp 154-155°C (hexane), $C_{20}H_{30}O_4Br_2$ [δ 5.32 (1H, dd, J=9.0, 3.5, 14-H), 2.91 (1H, ddd, J=16.0, 9.0, 6.5, 13β-H), 2.08 (1H, ddd, J=16.0, 10.0, 3.5, 13α-H), 2.17 (1H, m, 12-H)], and a bromo-hydrin (15), $C_{20}H_{31}O_5Br$ [δ 4.71 (1H, dd, J=8.0, 2.0, 14-H)], whereas OsO₄ oxidation of 5 yielded a triol (16), $C_{20}H_{32}O_6$, δ 3.92 (1H, m, 14-H).

¹H NMR examinations (500 MHz) of neodolabelline (5) and its derivatives (6-16), which included decoupling experiments, NOE examinations (Table I), and pyridine-induced solvent shifts⁷) (e.g. for 7-H in 8; for 5-H and 7-H in 12; for 2β -H and 7-H in 16), finally led us to presume a methyl migrated (C-1 \rightarrow C-11) dolabellane structure (5) for neodolabelline.

In order to substantiate the presumption and to establish the absolute stereochemistry of neodolabelline (5), the molecular structure with its absolute configuration of 1,14-dibromo-neodolabelline (14) was determined by the X-ray crystal structure analysis (Fig. 1).⁸⁾ Bond distances and angles were all normal. The C3, C4, C5, and C6 atoms were in a plane with the maximum deviation of 0.004 Å. The absolute structure of 5 was further confirmed by means of the cyclic α -glycol chirality method.⁹⁾ Thus, the CD curve [in CCl₄ with Eu(fod)₃] of *trans*-glycol (7) showed a positive Cotton effect: [θ]₃₀₅+19000 (pos. max.), while that of *cis*-glycol (8) showed only a weak Cotton curve: [θ]₃₃₁ -1500 (neg. max.), [θ]₃₀₂ +4000 (pos. max.). Consequently, the 8(R) and 9(S) configurations in 7 were confirmed.



Figure 1. ORTEP drawing of 1,14-dibromoneodolabelline (14). Thermal ellipsoids are drawn at the 50 % probability. Only three hydrogen atoms are shown to represent the absolute structure clearly.



16

	2 _{α-Н}	3-н	5-н	6β-н	7-н	14-н	18-н
5 ∼ NOE(%) ^{c)}	6[7-н]	10[2α-H] 8[2β-H] 6[4-CH3]	12[6α-H] 14[6β-H] 14[4-CH ₃]	15[6α-H] 5[8-CH ₃]	5[2a-H]	5[13α-H] 11[13β-H]	13[11-СН ₃] 13[18-СН ₃]
$NOE\left(\frac{7}{8}\right)c$		8[4-CH ₃]	12[4-СН ₃] 6[6α-Н]		· · · · · · · · · · · · · · · · · · ·		8[11-CH ₃]
<u>8</u> NOE (१) с)		8[4-CH ₃] 7[2β-H]	9[4-CH ₃]	5 [8-CH ₃]		4 [2β-н]	9[11-CH ₃]
$\frac{16}{(9)}c)$	7[11-CH ₃]	11[4-сн ₃]	8[4-cH ₃]		9[11-СН ₃]	10 [11-сн ₃]	6[11-СН ₃]

a) In d₆-DMSO solution. b) In CDCl₃ solution.

c) Saturated signals were shown in square brackets.

Various dolabellane-type diterpenes were isolated from sea hare, ¹⁰⁾ brown algae, 11 sea whip (sea fan), 12 and moss. 13 Neodolabelline (5) has a carbon skeleton, in which a CH₂ residue at C-1 in the dolabellane skeleton is migrated A few diterpenes such as 17 and 18, which have these methyl migrated to C-11. dolabellane skeletons, were reported so far from soft corals, Clavularia inflata¹⁴⁾ and Cespitularia sp.¹⁴⁾

Acknowledgement --- The authors are grateful to The Naito Foundation Research Grant for 1983 for financial support.

References and Notes

- 1) M. Kobayashi, Y. Kiyota, S. Orito, Y. Kyogoku, and I. Kitagawa, Tetrahedron Lett., in the press.
- 2) a) M. Kobayashi, B. W. Son, M. Kido, Y. Kyogoku, and I. Kitagawa, Chem. Pharm. Bull., 31, 2160 (1983); b) M. Kobayashi, B. W. Son, Y. Kyogoku, and I. Kitagawa, Chem. Pharm. Bull., 32, 1667 (1984).
 3) It should be mentioned here that the composition of clavukerins (1,2,3) and
- neodolabelline (5) in Clavularia koellikeri varied depending upon the year (1981-1983) collected, although the soft corals investigated were collected in the same waters and in the same season.
- 4) The molecular compositions of compounds with the chemical formulae were determined by high resolution mass spectrometry.
- 5) The spectra of 5, 6, 7, 8, 12, and 13 were measured in d₆-DMSO while those of 9, 10, 11, 14, 15, and 16 were taken in CDCl₃.
 6) B. W. Finucane and J. B. Thomson, Chem. Commun., 1969, 1220.
- 7) a) P. V. Demarco, E. Farkas, D. Doddrell, B. L. Mylari, and E. Wenkert, J. Am. Chem. Soc., 90, 5480 (1968); b) M. Kobayashi, T. Yasuzawa, Y. Kyogoku,
 M. Kido, and I. Kitagawa, Chem. Pharm. Bull., 30, 3431 (1982).
 A colorless crystal with dimensions of 0.16x0.34x0.25 mm was used for col-
- 8) lection of the X-ray diffraction data, and the color changed purple at the end of X-ray exposure. Two sets of intensity data were measured with the same crystal on a Rigaku AFC-5 diffractometer with the graphite monochromatized Cu-Ka radiation (1701 reflections, $2\theta \le 125^\circ$, and Fo $\ge 3\sigma$ Fo) and Mo-Ka (2201 hkl and hkl reflections, $2\theta \le 41.4^{\circ}$), the former was used for determination of the structure and the latter for that of the absolute structure, respectively. The crystallographic data were, $C_{20}H_{30}O_4Br_2$, $M_w=494.26$, monoclinic P2₁, a=9.328(3), b=15.233(3), c=7.395(2)Å, $\beta=99.66(2)^\circ$, V=1035.9 (5) Å³, D_x=1.585 gcm⁻³, Z=2, $\mu=56.98$ and 41.67 cm⁻¹ for Cu-K α and Mo-K α radiation, respectively. The structure was solved by the heavy atom method and refined by block-diagonal least-squares procedure with all hydrogen atoms except those on the methyl groups by setting them on their calculated posi-The final R and Rw values with 2201 Friedel pair reflections were tions. 0.045 and 0.053, respectively, whereas the enantiomeric structure refined under identical conditions gave R and Rw values of 0.063 and 0.072, respec-Tables of atomic coordinates, temperature factors, structure factively. tors, bond lengths and angles, planarity of atoms, have been deposited with the Cambridge Crystallographic Data Centre.
- 9) K. Nakanishi and J. Dillon, J. Am. Chem. Soc., <u>93</u>, 4058 (1971). 10) e.g. A. G. González, J. D. Martin, M. Norte, R. Pérez, V. Weyler, S. Rafii, and J. Clardy, Tetrahedron Lett., 24, 1075 (1983). 11) e.g. C. Tringali, M. Piattelli, and G. Nicolosi, Tetrahedron, 40, 799(1984).
- 12) S. A. Look and W. Fenical, J. Org. Chem., <u>47</u>, 4129 (1982). 13) A. Matsuo, K. Uohama, S. Hayashi, and J. D. Connolly, Chemistry Lett.,<u>1984</u>,
- 599.
- 14) B. F. Bowden, J. C. Braekman, J. C. Coll, and S. J. Mitchell, Aust. J. Chem. , 33, 927 (1980).

(Received in Japan 20 August 1984)